Utban Meteorology and Climate Conference

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Contribution and sources of light absorption brown carbon at a suburban site in Guangzhou, China

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Why care about Brown carbon?

- Quantitative prediction of radiative forcing due to aerosol is still challenging.
- Black carbon? Brown carbon (BrC)?
- BrC is an ensemble of light absorbing (colored) organic compounds.

- Particles from **smoldering combustion** can contain substantial amounts of BrC.
- Appears **light brown** to **yellowish**, and not black as would be expected for black carbon.



Sources and chemical compositions of BrC

Both **primary** and **secondary sources**:

- biomass burning, biological aerosols,
- secondary organic aerosol formed from anthropogenic or biogenic precursors.

Several classes of compounds:

- **nitroaromatic compounds**, such as nitrophenols
- N-heterocyclic compounds,
- and **quinones**.

BrC studies in Pearl River Delta region

- Light absorption of brown carbon aerosol in the PRD region of China by J.-F. Yuan, et al. Atmos. Chem. Phys., 2016
 BrC contributed 6.3% to 12.1% of the total aerosol light absorption at 405 nm; 4.1% to 10.0% at 532 nm.
- However, little is known about the relationship between the brown carbon and aerosol chemical compositions.

Sampling site

Sampling period: 7th Nov. 2014 - 3rd Jan. 2015



Panyu site:

- About 15 km south of the downtown Guanzhou
- Located at the summit of Dazhengang at ~150 m

Aethalometer (AE33):

Seven wavelengths aerosol light absorption

 $(\lambda = 370, 470, 520, 590, 660, 880 \text{ and } 950 \text{ nm})$



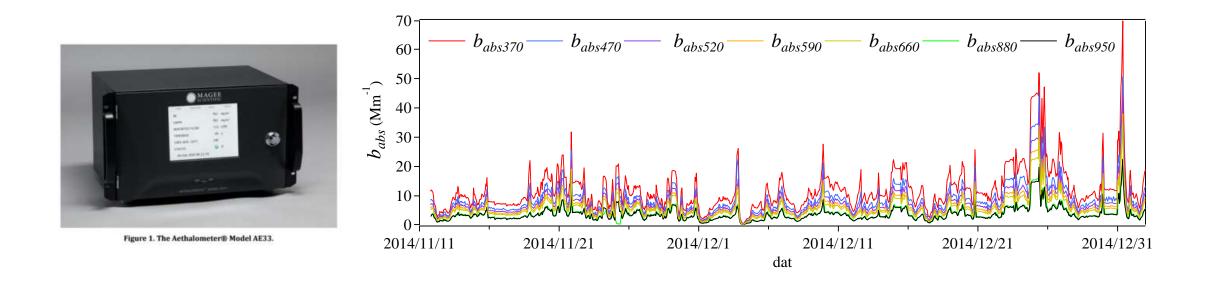
Figure 1. The Aethalometer® Model AE33.

High-resolution Time-of-Flight Aerosol Mass Spectrometer (HR-Tof-AMS):

Organics factors and organic nitrate, CHN,CHON CHOxN groups

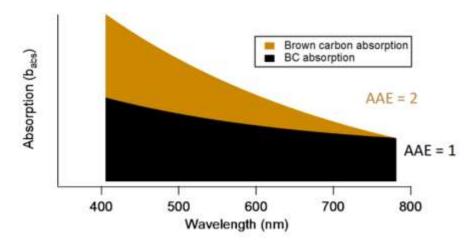


Time series of aerosol light absorption at different wavelengths



The average absorption was 11.8 Mm⁻¹ at 370nm , 8.6 Mm⁻¹ at 470 nm, 7.3 Mm⁻¹ at 520, 6.3 Mm⁻¹ at 590nm, 5.6 Mm⁻¹ at 660nm, 3.8 Mm⁻¹ at 880nm and 3.6 Mm⁻¹ at 950nm.

The basic theory used to describe light absorption properties is the well-known power-law relationship: $Abs_{\lambda} = K^* \lambda^{-AAE}$



where Abs_{λ} (unit: Mm⁻¹ or M⁻¹) refers to the absorption coefficient of aerosol at wavelength λ (unit: nm). K is a constant.

AAE indicates the wavelength dependence of light absorption.

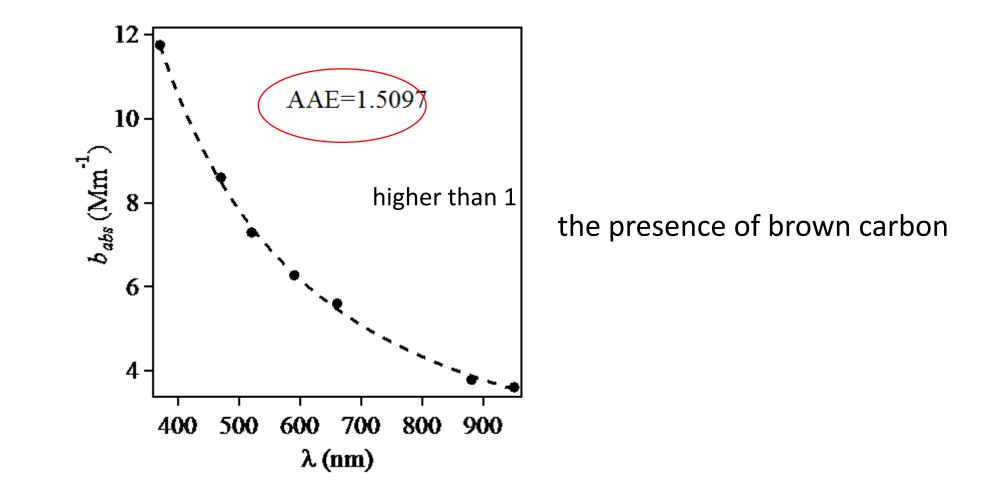
Pure black carbon: AAE ~1; AAE for brown carbon ≥ 2 , up to 9.7

The presence of BrC will make the absorption stronger at shorter wavelengths:

- BrC mainly absorb light at UV and short-visible wavelengths
- > and absorb negligibly at longer wavelength.

Average absorption spectra and AAE

$$Abs_{\lambda} = K^* \lambda^{-AAE}$$



Average absorption spectra over 370-950nm wavelength during the whole campaign.

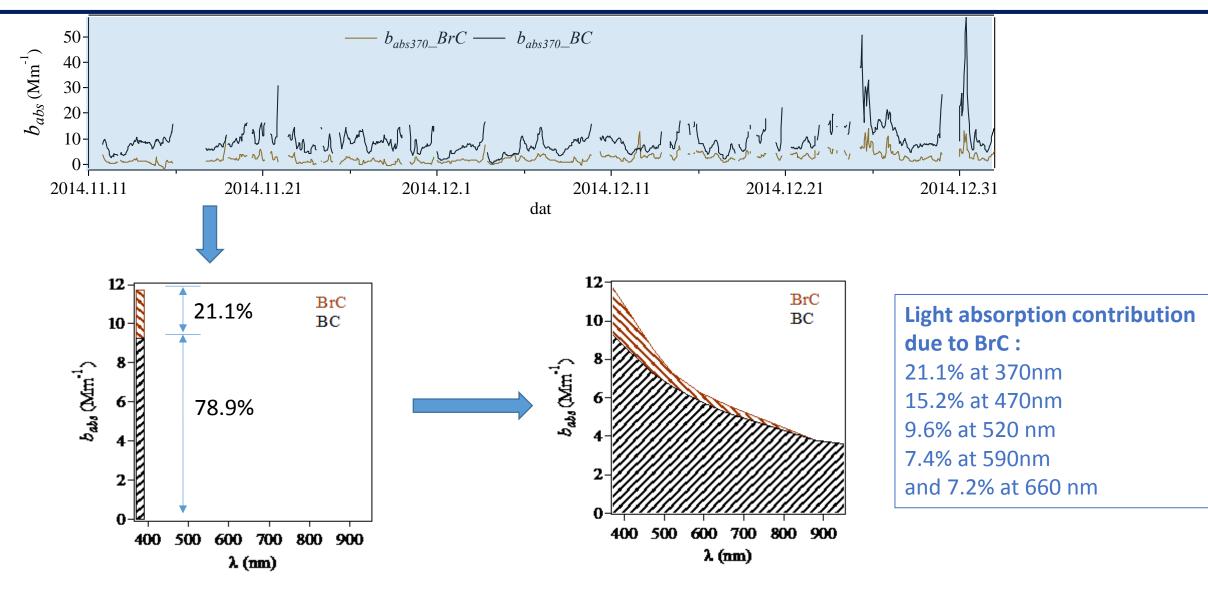
Attribution of aerosol light absorption to BC and BrC

- Power-law relationship $Abs_{\lambda} = K^* \lambda^{-AAE}$
- BrC mainly absorb light at UV and short-visible wavelengths and absorb negligibly at longer wavelength.

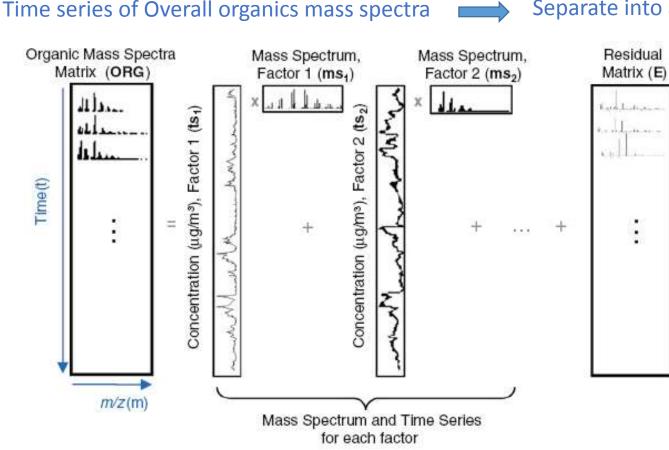
Theoretical AAE value for pure black carbon, "1"

Attributed absorption of BrC

Contribution of aerosol light absorption to BC and BrC



Organic factor analysis

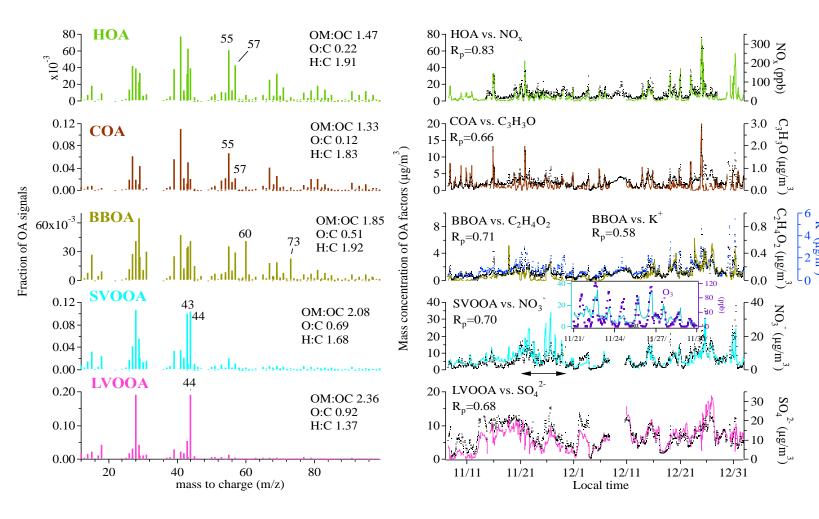


Separate into a number of factors

Each factor corresponds to a large group of OA constituents with similar chemical composition and temporal behavior.

 $ORG = TS \times MS + E$

Organic factor analysis



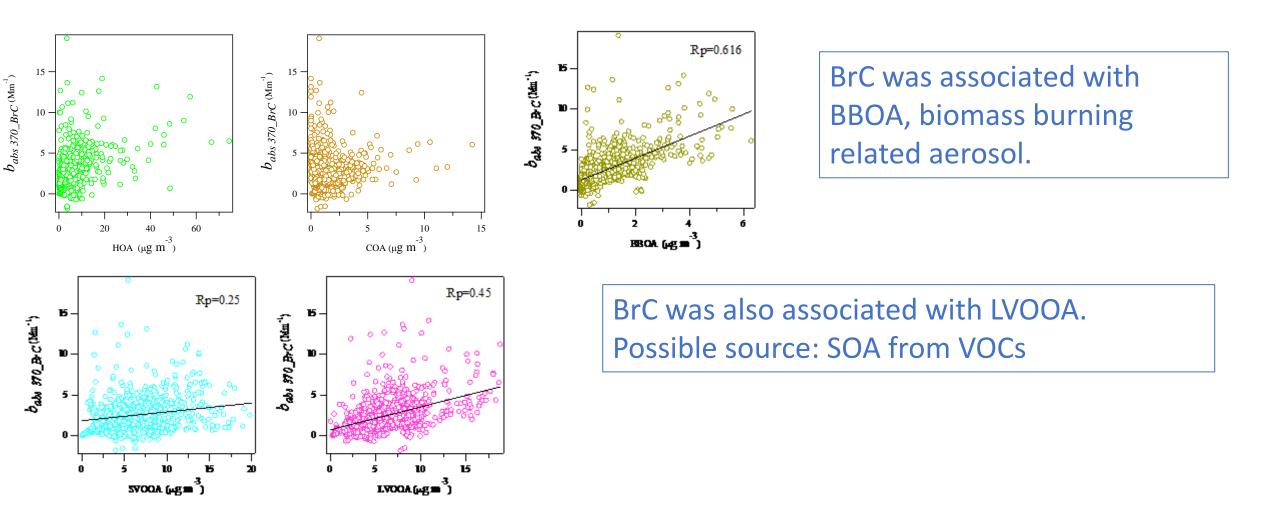
POA:

- Hydrocarbon-like organic aerosol (HOA);
- Cooking organic aerosol (COA);
- Biomass burning related organic aerosol (BBOA)

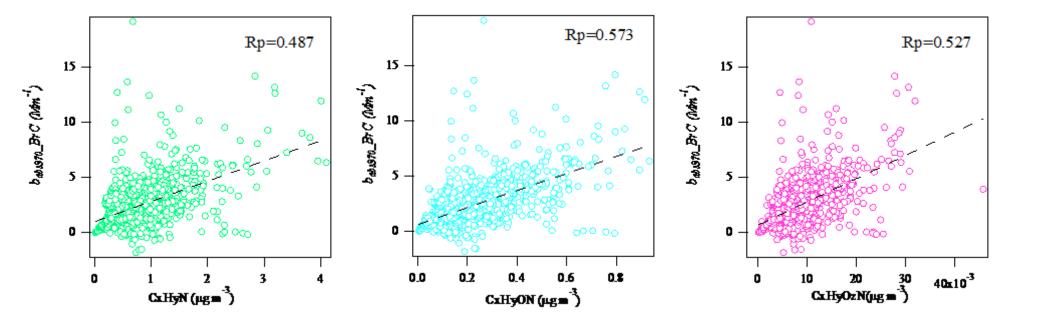
SOA:

- Semi-volatile oxygenated organic aerosol (SV-OOA);
- Low-volatility oxygenated organic aerosol (LVOOA)

Correlation of light absorption by BrC with OA



Correlation of light absorption by BrC with $C_xH_yN_rC_xH_yON$ and $C_xH_yO_xN$



 C_xH_yN , C_xH_yON and $C_xH_yO_xN$ may be responsible for the observed optical properties of BrC.



- I. Light absorption contribution due to BrC at Guangzhou, Panyu was obtained, with 21.1% of the total aerosol absorption at 370nm, 15.2% at 470nm, 9.6% at 520 nm, 7.4% at 590nm and 7.2% at 660 nm.
- II. The BrC were **associated with BBOA and LVOOA** at this site. Possible source: Biomass burning and SOA from VOCs.
- **III. CxHyN, CxHyON and CxHyOxN** may be responsible for the observed optical properties of BrC.



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